

(1*R*,2*R*)-(+)-1,2-Diphenylethylenediamine**Matthew D. Jones,* Filipe A. Almeida Paz, John E. Davies and Brian F. G. Johnson**

Department of Chemistry, University of Cambridge, Lensfield Road, Cambridge CB2 1EW, England

Correspondence e-mail: mdj22@cam.ac.uk

Key indicatorsSingle-crystal X-ray study
T = 180 K
Mean σ (C–C) = 0.003 Å
R factor = 0.035
wR factor = 0.082
Data-to-parameter ratio = 7.7For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The crystal structure of the title compound, $C_{14}H_{16}N_2$, has been determined at 180 (2) K in the non-centrosymmetric space group $P2_12_12_1$. The structure is formed by herring-bone-packed layers within which inter- and intramolecular N–H \cdots N hydrogen bonds can be found, forming a one-dimensional infinite chain. Intermolecular N–H $\cdots\pi$ interactions are also present.

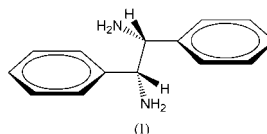
Received 21 February 2003

Accepted 6 March 2003

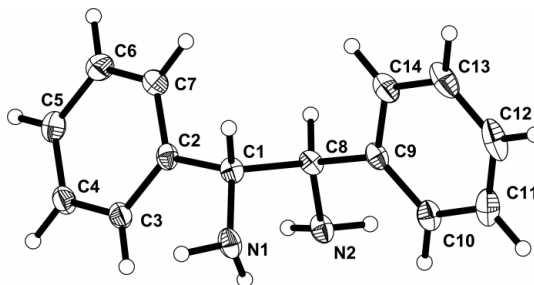
Online 14 March 2003

Comment

As part of our study devoted to the synthesis and characterization of novel chiral catalysts, we came across (1*R*,2*R*)-(+)-1,2-diphenylethylenediamine, (I), a bidentate chiral amine capable of forming chelates with transition metal cations (Jones *et al.*, 2003; Rouzaud *et al.*, 2003). Here we report the crystal structure, determined at 180 (2) K, of this chiral compound.



Compound (I) crystallizes in the orthorhombic non-centrosymmetric space group $P2_12_12_1$, with one whole molecule in the asymmetric unit (Fig. 1). Adjacent molecules of (I) are linked by a combination of inter- and intramolecular N–H \cdots N hydrogen bonds (Table 2 and Fig. 2), forming a one-dimensional infinite chain which runs along the **a** direction (Fig. 3). Further intermolecular N–H $\cdots\pi$ interactions are also evident in the crystal structure [$H2A\cdots Cg^i = 2.61(4)$ Å and $N2-H2A\cdots Cg^i = 138(4)^\circ$, where Cg is the centroid of the C9–C14 ring; symmetry code: (i) $x - 1, y, z$] (Fig. 2). Although one could expect to find a similar interaction

**Figure 1**

The molecular unit in (I), showing the labelling scheme for all non-H atoms. Displacement ellipsoids are drawn at the 30% probability level and H atoms are shown as small spheres.

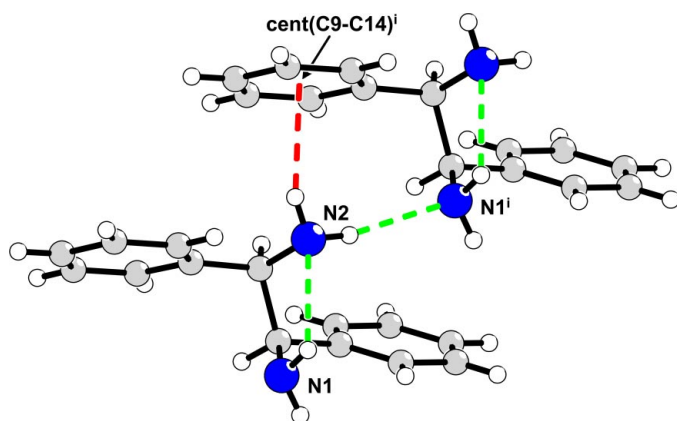


Figure 2
View, in detail, of the N—H...N intra- and intermolecular hydrogen bonds (dashed green lines) and the N—H... π interactions (red dashed lines) between adjacent molecules of (I). For hydrogen-bond details and symmetry code, see Table 2.

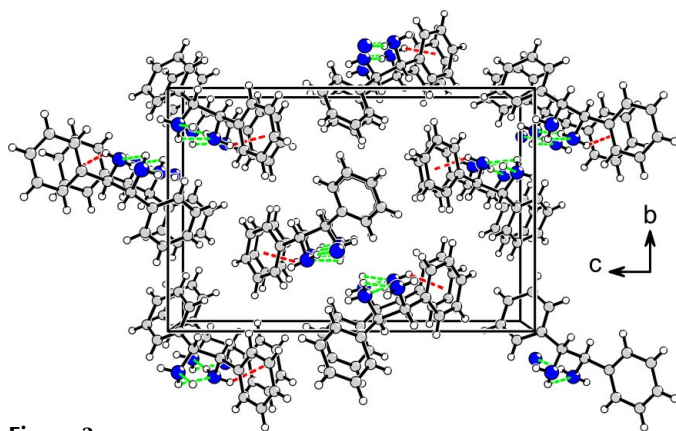


Figure 3
Perspective view of (I) along the **a** direction. Hydrogen bonds (see Table 2) are represented as green dashed lines, N—H... π interactions as red dashed lines.

between atom N1 and the neighbouring C2—C7 ring, the spatial arrangement of the molecules does not allow it. Individual molecules of (I) are stacked in a herring-bone manner, forming layers which pack in an *ABAB* fashion along the **b** direction (Fig. 3).

Experimental

(1*R*,2*R*)-(+)-1,2-Diphenylethylenediamine was purchased from Aldrich (99.99% purity) and used without further purification. Crystals suitable for X-ray diffraction analysis were obtained by recrystallization from methanol. ¹H NMR (CD₃OD): δ 3.95 (*s*, 1 H), 7.05–7.22 (*m*, 5 H). ¹³C NMR (CD₃OD): δ 64.2 (CH), 128.5, 128.8, 129.5, 144.2 (Ph).

Crystal data

C ₁₄ H ₁₆ N ₂	Mo <i>K</i> α radiation
<i>M_r</i> = 212.29	Cell parameters from 4851 reflections
Orthorhombic, <i>P</i> 2 ₁ 2 ₁ 2 ₁	θ = 1.0–25.0°
<i>a</i> = 5.1447 (1) Å	μ = 0.07 mm ^{−1}
<i>b</i> = 12.3264 (4) Å	<i>T</i> = 180 (2) K
<i>c</i> = 18.5357 (7) Å	Block, colourless
<i>V</i> = 1175.45 (6) Å ³	0.23 × 0.18 × 0.12 mm
<i>Z</i> = 4	
<i>D_x</i> = 1.200 Mg m ^{−3}	

Data collection

Nonius KappaCCD diffractometer	1148 reflections with <i>I</i> > 2 σ (<i>I</i>)
Thin-slice ω and φ scans	<i>R</i> _{int} = 0.056
Absorption correction: multi-scan (SORTAV; Blessing, 1995)	θ_{\max} = 25.0°
<i>T</i> _{min} = 0.895, <i>T</i> _{max} = 0.992	<i>h</i> = −6 → 6
6085 measured reflections	<i>k</i> = −12 → 14
1229 independent reflections	<i>l</i> = −21 → 22

Refinement

Refinement on <i>F</i> ²	$w = 1/[\sigma^2(F_o^2) + (0.0337P)^2 + 0.2099P]$
<i>R</i> [<i>F</i> ² > 2 σ (<i>F</i> ²)] = 0.035	where <i>P</i> = (<i>F</i> _o ² + 2 <i>F</i> _c ²)/3
<i>wR</i> (<i>F</i> ²) = 0.082	(Δ/σ) _{max} < 0.001
<i>S</i> = 1.09	$\Delta\rho_{\max}$ = 0.14 e Å ^{−3}
1229 reflections	$\Delta\rho_{\min}$ = −0.14 e Å ^{−3}
159 parameters	Extinction correction: <i>SHELXL97</i>
H atoms treated by a mixture of independent and constrained refinement	Extinction coefficient: 0.038 (6)

Table 1

Selected geometric parameters (Å, °).

N1—C1	1.456 (3)	C6—C7	1.380 (3)
N2—C8	1.463 (3)	C8—C9	1.516 (3)
C1—C2	1.524 (3)	C9—C10	1.387 (3)
C1—C8	1.551 (2)	C9—C14	1.394 (3)
C2—C3	1.390 (3)	C10—C11	1.390 (3)
C2—C7	1.394 (3)	C11—C12	1.378 (4)
C3—C4	1.390 (3)	C12—C13	1.372 (4)
C4—C5	1.379 (3)	C13—C14	1.385 (3)
C5—C6	1.392 (3)		
N1—C1—C2	116.33 (15)	N2—C8—C9	111.32 (16)
N1—C1—C8	108.93 (16)	N2—C8—C1	108.94 (15)
C2—C1—C8	109.41 (16)	C9—C8—C1	111.57 (16)
C3—C2—C1	117.77 (19)	C10—C9—C14	117.90 (19)
C3—C2—C7	123.08 (18)	C10—C9—C8	122.91 (18)
C7—C2—C1	119.13 (17)	C14—C9—C8	119.19 (18)
C4—C3—C2	120.46 (19)	C9—C10—C11	120.7 (2)
C5—C4—C3	121.18 (19)	C12—C11—C10	120.4 (2)
C4—C5—C6	118.9 (2)	C13—C12—C11	119.7 (2)
C7—C6—C5	119.8 (2)	C12—C13—C14	120.0 (2)
C6—C7—C2	121.85 (19)	C13—C14—C9	121.2 (2)

Table 2

Hydrogen-bonding geometry (Å, °).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N1—H1 <i>B</i> ...N2	0.916 (17)	2.34 (2)	2.832 (2)	113.1 (16)
N2—H2 <i>B</i> ...N1 ⁱ	0.927 (18)	2.346 (18)	3.249 (2)	164.3 (17)

Symmetry code: (i) *x* − 1, *y*, *z*.

H atoms bound to C atoms were placed in calculated positions and allowed to ride during subsequent refinement, with *U*_{iso}(H) = 1.2*U*_{eq}(C). The NH₂ H atoms were located from difference Fourier maps and refined successfully. A total of 801 Friedel pairs were merged and not used as independent data. The corresponding Flack (1983) parameter was found to be meaningless and was omitted.

Data collection: *COLLECT* (Nonius, 1998); cell refinement: *HKL SCALEPACK* (Otwinowski & Minor, 1997); data reduction: *HKL DENZO* (Otwinowski & Minor, 1997) and *SCALEPACK*; program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *SHELXTL* (Bruker, 2001); molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

We thank the EPSRC for a studentship to MDJ and for their general financial support, ICI for financial support, and the

Newton Trust. We are also grateful to the Portuguese Foundation for Science and Technology (FCT) for financial support through the PhD scholarship No. SFRH/BD/3024/2000 given to FAAP.

References

- Altomare, A., Cascarano, G., Giacovazzo, C., Guagliardi, A., Burla, M. C., Polidori, G. & Camalli, M. (1994). *J. Appl. Cryst.* **27**, 435.
- Blessing, R. H. (1995). *Acta Cryst.* **A51**, 33–38.
- Bruker (2001). *SHELXTL*. Version 6.12. Bruker AXS Inc., Madison, Wisconsin, USA.
- Flack, H. D. (1983). *Acta Cryst.* **A39**, 876–881.
- Jones, M. D., Raja, R., Thomas, J. M., Johnson, B. F. G., Lewis, D. W., Rouzaud, J. & Harris, K. D. M. (2003). *Angew. Chem. Int. Ed.* Submitted.
- Nonius (1998). *COLLECT*. Nonius BV, Delft, The Netherlands.
- Otwinowski, Z. & Minor, W. (1997). *Methods in Enzymology*, Vol. 276, *Macromolecular Crystallography*, Part A, edited by C. W. Carter Jr and R. M. Sweet, pp. 307–326. New York: Academic Press.
- Rouzaud, J., Jones, M. D., Raja, R., Johnson, B. F. G., Thomas, J. M. & Duer, M. J. (2003). *Chem. Commun.* Submitted.